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CHAPTER 3 PRESSURE AND FLUID STATICS

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SOLUTION $p=0$. The fluid is considered compressible. $dp = \rho V dv$. However, $V = dV > V$. m . Then, $r = h z - (m > r^2)$ $dp = V$. $dr = -m > r$. r . Therefore, $EV = p$. $dp dr > r$ (a) At the surface, where $p \dots$

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Solution: $p_{abs} = p_{atm} + p_{gage} = 101.3 \text{ kPa} + 20 \text{ in. H}_2\text{O}$
 $1.248 \text{ Pa} = 110.26 \text{ kPa}$ Ideal gas law: $\rho = \frac{p}{RT} = \frac{110.26 \text{ kPa}}{1 \text{ kg/K} \cdot 2077 \text{ J/K} \cdot 293.2 \text{ K}} = 1000 \text{ Pa} / 1 \text{ kg/m}^3$
 $\rho = 0.181 \text{ kg/m}^3$ b.) Situation: A sphere of 93 mm diameter contains an ideal gas. $T = 20 \text{ C} = 293.2 \text{ K}$ Find: Calculate the density of argon at a vacuum pressure of 8.8 psi. Properties:

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